

## **Course Specifications**

Course Title:	Chemistry of Natural Products
Course Code:	CHEM 436
Program:	Bachelor in Chemistry
Department:	Chemistry
College:	Science
Institution:	Jazan University (JU)









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## A. Course Identification

1. (	Credit hours:	3 hrs.	Workload:	173	ECTS: 6.2
2. 0	Course type				
a.	University	College	Departi	ment $$	Others
b.	Requ	uired $$	Elective		
3. 1	Level/year at w	hich this cou	irse is offered:	:	
				7 <sup>th</sup> lev	vel /Fourth year
4. ]	Pre-requisites f	or this cours	se (if any):		
			No	one	
5. (	Co-requisites fo	or this cours	e (if any):		
			No	one	

## **6. Mode of Instruction** (mark all that apply)

No	Mode of Instruction	Contact Hours	Percentage
1	Traditional classroom and Lab	30 and 30	100%
2	Blended		
3	E-learning		
4	Distance learning		
5	Other		

#### 7. Contact Hours (based on academic semester)

No	Activity	<b>Contact Hours</b>
1	Lecture	30
2	Laboratory/Studio	30
3	Tutorial	0
4	Others (specify)	0
	Total	60

## **B.** Course Objectives and Learning Outcomes

#### 1. Course Description

Ľ											
	Course Title	Course Number	Contact Hours (CH)				Credit unit	Year	Level	Pre- requisite	
	ļ		Lec.	Prac.	( <b>CU</b> )			requisite			
	Chemistry of	CHEM 436	2	2	3	Fourth	7th	None			
	Natural Products						/				

**Course objectives:** They are to identify the following.

- 1. The main classes of natural products and their types.
- 2. Terpenoid; its classification and methods of isolation.
- 3. Amino and fatty acids, alkaloids, their importance, types and means of extraction.
- 4. Natural phenolic compounds and their extraction, isolation and structure elucidation

#### **Syllabus: A-Theoretical contents**

Definition and classification of different classes of natural products and their isolation by different chromatographic methods- Structure elucidation by means of physical and chemical methods. Some chemical reactions and biosynthesis of terpenes, steroids, alkaloids and natural phenolic (flavonoids, xanthones, anthraquinons and coumarins

#### Syllabus: B-Practical contents

Preparation and identification of some organic compounds, (such as aspirin - Benzoyl Glycine – benzamide - phthalimide - picric acid - P- nitro-acetanilide, etc.)

\*See attachment

#### 2. Course Main Objective

This course aims to provide students with the basic knowledge about the main classes of natural products, means of extraction, isolation, structure characterization, and their most important uses

	CLOs	Aligned PLOs
1	Knowledge and Understanding	
	Upon completion of the course, student will be able to:	
1.1	Demonstrate broad and understanding knowledge about different classes of natural products, their importance and identification, the physical and chemical properties of different secondary metabolites. (M)	K1
1.2	Describe the chromatographic procedures that used in extraction and isolation of each class of natural products and the ways for structure elucidation of simple natural secondary metabolites (M)	K2
2	Shille .	
	Skills :	
	<i>Upon completion of the course, student will be able to:</i>	
2.1		S1
2.1 2.2	Upon completion of the course, student will be able to: Demonstrate critical thinking ability to isolated compounds using different structure	\$1 \$2
	Upon completion of the course, student will be able to:         Demonstrate critical thinking ability to isolated compounds using different structure elucidation methods       (P)         Apply their experimental basics and skills to use laboratory equipment, modern instructions, and classical techniques to Perform experiments to prepare a simple	

3	Values:		
	Upon completion of the course, student will be able to:		
3.1	Working as a group leader in cooperation with other colleagues.	(P)	V1

## **C.** Course Content

No	No List of Topics			
1	Introduction to natural products, classification, general methods of extractions, isolation and structure elucidation	4		
2	Classification and identification of terpenoids, methods of extraction, chemical properties and structure elucidation	8		
3	Classification and identification of alkaloids, methods of extraction, chemical properties and structure elucidation	8		
4	Classification and identification of phenolics, isolation and chemical properties	5		
5	Miscellaneous natural products	5		
6	Selected experiments on preparation, isolation and purification of simple organic compounds	30		
	Total	60		

## **D.** Teaching and Assessment

# **1.** Alignment of Course Learning Outcomes with Teaching Strategies and Assessment Methods

Code	Course Learning Outcomes	Teaching Strategies	Assessment Methods
1.0	Knowledge and Understanding		
1.1	Demonstrate broad and understanding knowledge about different classes of natural products, their importance and identification, the physical and chemical properties of different secondary metabolites. ( M)	Lectures, directed reading, group discussion and assignments	MCQ and short answer questions
1.2	Describe the chromatographic procedures that used in extraction and isolation of each class of natural products and the ways for structure elucidation of simple natural secondary metabolites (M)	Lectures, directed reading, group discussion and assignments	MCQ and short answer questions
2.0	Skills		
2.1	Demonstrate critical thinking ability to isolated compounds using different structure elucidation methods (P)	Lectures, directed reading, group discussion and assignments	Written, questions, problems and class discussions
2.2	Apply their experimental basics and skills to use laboratory equipment, modern instructions, and classical techniques to Perform experiments to prepare a simple organic compound (M)	Lab work, group work	Practical sheet and final exam
2.3	Examine and follow proper procedures and regulations for safe handling, use, and disposal of chemicals. (P)	Lab work	MCQ safety exam
	Make effective use of communication, modern library searching and information technology about	Group Discussion and research	Presentation and assessment research

	Natural	products	Ι				
	(I)						
3.0	Values						
2.1	Working as a group le	ader in cooperation with other	Group	work	and	Presentation	and
3.1	colleagues.	(P)	projects			assessment resear	ch

#### 2. Assessment Tasks for Students

#	Assessment task*	Week Due	Percentage of Total Assessment Score
1	Homework assignment	3	1%
2	Quiz 1	5	2%
3	Oral presentation	7	0%
4	Homework assignment	8	1%
5	Mid-term exam	9	15%
6	Research project report	11	1%
7	Quiz in lab safety	14	0%
8	Final practical exam	15	20%
9	Final examination sheet in the practical part	15	10%
10	Final Exam	16	50%

\*Assessment task (i.e., written test, oral test, oral presentation, group project, essay, etc.)

## E. Student Academic Counseling and Support

Arrangements for availability of faculty and teaching staff for individual student consultations and academic advice:

Student Academic Counseling:

Members of staff will be available for academic counseling on daily basis for at 4h/day during office hours.

## **F. Learning Resources and Facilities**

## **1. Learning Resources**

Required Textbooks	كيمياء المنتجات الطبيعية – الجزء النظري, أ.د. طاهر حسن, جامعة البعث, مديرية الكتب المطبو عات الجامعية Chemistry of Natural Products, S. V. Bhat, B. A. Nagasampagi, S. Minakshi, Springer, 2005		
Essential References Materials	I - Natural Products Isolation, S. D. Saker, Z. Latif, A. I. Gray, 2 <sup>nd</sup> ed., Humana Pre-		
Electronic Materials	https://chem.libretexts.org/Special:Search?qid=&fpid=230&fpth=&query=Natural+prod ucts&type=wiki https://chem.libretexts.org/Bookshelves/Organic Chemistry/Book%3A Basic Principle s_of_Organic_Chemistry_(Roberts_and_Caserio)/30%3A_Natural_Products_and_Biosy nthesis		
Other Learning Materials	www.wikipedia.org https://www.slideshare.net/ShvetaArya/chemistry-of-naturalproducts		

#### 2. Facilities Required

Item	Resources
Accommodation	1 Lecture room (s) for each group of 25 students.
(Classrooms, laboratories, demonstration	One Laboratory for each group of 15 students.

rooms/labs, etc.)		
<b>Technology Resources</b> (AV, data show, Smart Board, software, etc.)	AV, data show, Smart Board, software, etc.)	
<b>Other Resources</b> (Specify, e.g. if specific laboratory equipment is required, list requirements or attach a list)	In addition to glassware and chemicals some equipment are required for teaching the practical part such as: 1- UV/Vis Spectrophotometer 2- IR spectrophotometer 3- NMR spectrometer 4- Polarimeter	

#### **G.** Course Quality Evaluation

Evaluation Areas/Issues	Evaluators	<b>Evaluation Methods</b>
Effectiveness of Teaching and Assessment	Student	Likert-type Survey (CES) Indirect
Extent of achievement of course learning outcomes	Instructor & Course coordinator	Class room evaluation (direct & indirect)
Quality of learning resources	Program coordinator	Indirect
Exam Quality assessment	Assessment committee	Indirect

**Evaluation areas** (e.g., Effectiveness of teaching and assessment, Extent of achievement of course learning outcomes, Quality of learning resources, etc.)

**Evaluators** (Students, Faculty, Program Leaders, Peer Reviewer, Others (specify) Assessment Methods (Direct, Indirect)

## H. Specification Approval Data

Council / Committee	Chemistry Department Council	
Reference No.	42 / 35 /102 112	
Date	17 /09 /1442 Corresponding to 28 / 04 /2021	

	CHEM 436 LAB			
	Experiment	Equipments, Chemicals and Tools	Week Due	Remarks
1-	Lab Safety		First week	The required equipment and tools for teaching the
2-	Preparation of Aspirin	Salicylic acid, acetic anhydride	Second week	
3-	Preparation of acetanilide	Aniline acetic anhydride	Third week	<ul> <li>practical part are:</li> <li>UV/Vis</li> <li>Spectrophotometer</li> </ul>
4-	Preparation of p- nitroacetanilide	Acetanilide , Nitric acid, and Sulphuric acid	Fourth week	<ul> <li>Spectrophotometer</li> <li>IR</li> <li>spectrophotometer</li> <li>NMR spectrometer</li> <li>Polarimeter</li> <li>Hotplate magnetic stirrer</li> <li>Condensers</li> <li>Separating funnels different sizes</li> <li>Rotary evaporator</li> <li>Melting point apparatus</li> <li>Heating mantle</li> <li>TLC sheets</li> <li>Capillary tubes</li> </ul>
5-	Preparation of Phthalimide	Phthatic anhydride and Urea	Fifth week	
6-	Preparation of Phthalyl glycine	Phthatic anhydride and Glycine	Sixth week	
7-	Preparation of benzoin	Benzaldehyde and Potassium cyanide	Seventh week	
8-	Preparation of benzophenone Oxime	Benzophenone and hydroxylamine hydrochloride	Eighteenth week	
9-	Preparation of 7- hydroxycoumarine	Resorcinol, ethyl acetoacetate, and sulphuric acid	Ninth week	
10-	Extraction of caffeine from green tea	Green tea, chloroform, and separating funnel	Tenth and Eleventh week	
11-	IR spectra of selected prepared compounds	Infra-Red Spectroscopy apparatus	Twelfth week and Thirteenth week	
12-	Revision		Fourteenth week	
13-	Quiz and Final exam		Fifteenth week	

CHEM 136 LAB

